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BOOK OF ABSTRACTS



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INVESTIGATION OF PHOTODEPOSITED PLATINUM LOCALIZATION AT THE LAYERED OXIDES HCa₂Nb₃O₁₀, H₂La₂Ti₃O₁₀ AND THEY ORGANO-INORGANIC HYBRIDS

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In recent years, environmental problems caused by using fossil fuel have become particularly acute. Hydrogen produced by the photocatalytic splitting of water and organic substrates under the action of sunlight can be a promising replacement for hydrocarbon fuel. Of particular interest among heterogeneous photocatalysts is a class of layered perovskite-like oxides, in the structure of which there is an alternation of negatively charged perovskite blocks with blocks of a different structure, called interlayer space, which can act as a separate reaction zone in the photocatalytic process [1]. The introduction of organic molecules, for example amines, into interlayer space contributes to an increase in the photocatalytic activity of samples in the reaction of obtaining hydrogen from a 1% (mol.) agueous solution of methanol under the action of ultraviolet irradiation [2]. In this case, samples additionally modified with photo-deposited platinum particles exhibit particularly high activity. It was assumed that this effect is caused by the expansion of the interlayer space, which led to an increase in the accessibility of this zone for water and methanol molecules. The organic components undergo various transformations due to irradiation with ultraviolet light [3]. At the same time the activity of the photocatalyst remains constant throughout the experiment. A possible reason for the immutability of the activity of organo-inorganic hybrids may also be the deposition of platinum particles in the interlayer space expanded by the introduction of organic matter. In this instance, the activity of the samples would be preserved even after the decomposition of the organic modifiers.

This work focuses on the study of the effect of organic modifiers on the process of photodeposition of platinum particles and the relationship of this process with the observed photocatalytic activity of hybrid samples of the layered oxides HCa2Nb3O10 and H2La2Ti3O10. It was found that the sequence of procedures for modifying layered oxide H2La2Ti3O10 by intercalation of alkylamines RNH2 (R = Me, Oc) and platination significantly affects the stability of the formed platinum particles to the action of aqua regia. The photocatalytic activity of organic platinized derivatives in the reaction of hydrogen extraction from a 1% (mol.) aqueous methanol solution under UV irradiation was also compared. It was shown that all modified samples exceed the initial H2La2Ti3O10/Pt in terms of photocatalytic activity by 3 times or more, but they did not differ so markedly from each other. As the result it was found that intercalation of organic molecules can affect the deposition of platinum particles in layered oxide. At the same time, the stable photocatalytic activity of the samples is not due to the different spatial distribution of Pt nanoparticles. Thus, the increased activity of photocatalysts is rather associated with the modification of oxides by organic molecules.

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