# New arsenate minerals from the Arsenatnaya fumarole, Tolbachik volcano, Kamchatka, Russia. XX. Evseevite, Na<sub>2</sub>Mg(AsO<sub>4</sub>)F, the first natural arsenate with antiperovskite structure

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Running title: Evseevite, a new mineral

# Abstract

The new mineral evseevite was found in the Arsenatnaya fumarole, Second scoria cone of the Northern Breakthrough of the Great Tolbachik Fissure Eruption, Tolbachik volcano, Kamchatka, Russia. Evseevite is represented by two chemical varieties. The variety close to the end-member Na<sub>2</sub>Mg(AsO<sub>4</sub>)F (holotype) is associated with sanidine, hematite, tenorite, aegirine, cassiterite, sylvite, halite, johillerite, badalovite, calciojohillerite, hatertite, arsmirandite, yurmarinite, axelite, polyarsite, aphthitalite, potassic-magnesio-fluoro-arfvedsonite, litidionite, ferrisanidine,



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and tridymite. The P- and S-enriched variety (cotype) is associated with hematite, fluorophlogopite, svabite, fluorapatite, tilasite, calciojohillerite, forsterite, cassiterite, belomarinaite, and aphthitalite. Evseevite occurs as prismatic, acicular or hair-like crystals up to 0.7 mm long combined in clusters up to 0.5 mm, brushes or crusts up to  $2 \times 2$  mm. It is transparent, colourless or pale pinkish, with vitreous lustre.  $D_{calc}$  is 3.377 g cm<sup>-3</sup> for holotype and 3.226 g cm<sup>-3</sup> for cotype. Evseevite is optically uniaxial (+),  $\alpha$  1.545(2),  $\beta$  1.546(2),  $\gamma$  1.549(2), and  $2V_{\text{meas}}$  40(10)°. The empirical formulae calculated based on O+F = 5 apfu are  $(Na_{1.99}Ca_{0.03}K_{0.01})_{\Sigma_{2.03}}(Mg_{0.98}Fe^{3+}_{0.01}Zn_{0.01}Cu_{0.01})_{\Sigma_{1.01}}[(As_{0.98}Si_{0.01})_{\Sigma_{1.01}}O_4](F_{0.97}O_{0.03})$  for holotype and Na<sub>2.02</sub>(Mg<sub>1.00</sub>Fe<sup>3+</sup> $_{0.03}$ )<sub> $\Sigma$ 1.03</sub>[(As<sub>0.69</sub>P<sub>0.25</sub>S<sub>0.07</sub>)<sub> $\Sigma$ 1.01</sub>O<sub>4</sub>](F<sub>0.78</sub>O<sub>0.22</sub>) for cotype. Evseevite is orthorhombic, *Pbcn*, a 5.3224(1), b 14.1255(3), c 12.0047(3) Å, V 902.53(4) Å<sup>3</sup> and Z = 8. Strong reflections of the powder XRD pattern [d, A(I)(hkl)] are: 4.001(100)(121), 3.479(56)(023), 3.041(45)(042), 2.657(44)(200), 2.642(68)(142) and 2.613(36)(104). The crystal structure was solved from single-crystal XRD data and refined on powder data by Rietveld method,  $R_{wp} = 0.0068$ ,  $R_p = 0.0047$ ,  $R_{obs} = 0.0435$ . Evseevite is isostructural to moraskoite Na<sub>2</sub>Mg(PO<sub>4</sub>)F. The structure of evseevite can be described in terms of anion-centred polyhedra. F-centred octahedra [FNa<sub>4</sub>Mg<sub>2</sub>]<sup>7+</sup> share faces to form chains [FNa<sub>2</sub>Mg]<sup>3+</sup> and AsO<sub>4</sub> tetrahedra are located between the chains. Evseevite belongs to quite few minerals with antiperovskite structures being the first natural arsenate with antiperovskite units. The mineral is named in honor of the Russian mineralogist Aleksandr Andreevich Evseev (born 1949).

**Keywords**: evseevite; new mineral; sodium magnesium fluorarsenate; crystal structure; anioncentred octahedron; antiperovskite structure; fumarole sublimate; Tolbachik volcano.

## Introduction

This paper continues the series of articles in which we characterize new arsenates from the Arsenatnaya fumarole at the Second scoria cone of the Northern Breakthrough of the Great Tolbachik Fissure Eruption 1975–1976 (NB GTFE), Tolbachik volcano, Kamchatka Peninsula, Far-Eastern Region, Russia. Twenty two new mineral species were described in the previous papers of the series: yurmarinite Na<sub>7</sub>(Fe<sup>3+</sup>,Mg,Cu)<sub>4</sub>(AsO<sub>4</sub>)<sub>6</sub> (Pekov *et al.*, 2014a), two polymorphs of Cu<sub>4</sub>O(AsO<sub>4</sub>)<sub>2</sub>, ericlaxmanite and kozyrevskite (Pekov *et al.*, 2014b), popovite Cu<sub>5</sub>O<sub>2</sub>(AsO<sub>4</sub>)<sub>2</sub> (Pekov *et al.*, 2015a), structurally related shchurovskyite K<sub>2</sub>CaCu<sub>6</sub>O<sub>2</sub>(AsO<sub>4</sub>)<sub>4</sub> and dmisokolovite K<sub>3</sub>Cu<sub>5</sub>AlO<sub>2</sub>(AsO<sub>4</sub>)<sub>4</sub> (Pekov *et al.*, 2015b), katiarsite KTiO(AsO<sub>4</sub>) (Pekov *et al.*, 2016a), melanarsite K<sub>3</sub>Cu<sub>7</sub>Fe<sup>3+</sup>O<sub>4</sub>(AsO<sub>4</sub>)<sub>4</sub> (Pekov *et al.*, 2018b), arsenatrotitanite NaTiO(AsO<sub>4</sub>) (Pekov *et al.*, 2019a), the two isostructural minerals edtollite K<sub>2</sub>NaCu<sub>5</sub>Fe<sup>3+</sup>O<sub>2</sub>(AsO<sub>4</sub>)<sub>4</sub> and

alumoedtollite  $K_2NaCu_5AlO_2(AsO_4)_4$ (Pekov al., 2019b), anatolyite et Na<sub>6</sub>(Ca,Na)(Mg,Fe<sup>3+</sup>)<sub>3</sub>Al(AsO<sub>4</sub>)<sub>6</sub> (Pekov *et al.*, 2019c), zubkovaite Ca<sub>3</sub>Cu<sub>3</sub>(AsO<sub>4</sub>)<sub>4</sub> (Pekov *et al.*,  $K_{3}Na_{3}Fe^{3+}(AsO_{4})_{8}$ (Pekov 2019d), pansnerite et al., 2020a), badalovite NaNaMg(MgFe<sup>3+</sup>)(AsO<sub>4</sub>)<sub>3</sub> (Pekov et al., 2020b), calciojohillerite NaCaMgMg<sub>2</sub>(AsO<sub>4</sub>)<sub>3</sub> (Pekov et al., 2021a), yurgensonite K<sub>2</sub>SnTiO<sub>2</sub>(AsO<sub>4</sub>)<sub>2</sub> (Pekov *et al.*, 2021b), paraberzeliite NaCaCaMg<sub>2</sub>(AsO<sub>4</sub>)<sub>3</sub> (Pekov et al., 2022a), khrenovite Na<sub>3</sub>Fe<sup>3+</sup><sub>2</sub>(AsO<sub>4</sub>)<sub>3</sub> (Pekov et al., 2022b), and axelite Na<sub>14</sub>Cu<sub>7</sub>(AsO<sub>4</sub>)<sub>8</sub>F<sub>2</sub>Cl<sub>2</sub> (Pekov *et al.*, 2023).

In this article we describe the new mineral evseevite (Cyrillic: евсеевит), ideally Na<sub>2</sub>Mg(AsO<sub>4</sub>)F. It is named in honour of the Russian mineralogist Aleksandr Andreevich Evseev (born 1949) who works in Fersman Mineralogical Museum of the Russian Academy of Sciences, Moscow. He is a specialist in history of mineralogy and an application of geographical information in mineralogy.

Both the new mineral and its name have been approved by the IMA Commission on New Minerals, Nomenclature and Classification (IMA CNMNC), IMA2019–064. The holotype specimen is deposited in the systematic collection of the Fersman Mineralogical Museum, Moscow with the catalogue number 96701.

# Occurrence and general appearance

The Arsenatnaya fumarole, discovered by us in 2012, is located at the summit of the Second scoria cone of the NB GTFE, a monogenetic volcano formed in 1975 (Fedotov and Markhinin, 1983). Arsenatnaya is one of the hottest Tolbachik fumaroles today: the temperatures measured by us using a chromel-alumel thermocouple in the period 2012–2022, reach 500°C in its deep levels. It is a locality greatly outstanding in mineral diversity and originality: more than 200 mineral species have been reliably identified here, including 67 new minerals. The mineralogy and zonation of sublimate incrustations of Arsenatnaya were recently described by Pekov *et al.* (2018a) and Shchipalkina *et al.* (2020); the general mineralogical features of oxidizing-type fumaroles of Tolbachik were overviewed by Vergasova and Filatov (2016) and Pekov *et al.* (2020c).

The specimens with the new mineral were first collected by us in July 2018 from hot pockets situated 1.5-2 m below the day surface, within the zone IV (Shchipalkina *et al.*, 2020), or polymineralic zone. Evseevite is represented by two chemical varieties. The variety close to the end-member Na<sub>2</sub>Mg(AsO<sub>4</sub>)F gave the specimen considered as the holotype (sample #6328<sup>1</sup>): all complex of studies was carried out for this variety, including the crystal structure determination.

<sup>&</sup>lt;sup>1</sup> ##6328, 6260 and 7529 are authors' working numbers of the three studied samples; #6328 is corresponding to the catalogue number 96701 in the systematic collection of the Fersman Mineralogical Museum (see Introduction).

The variety enriched with admixed P and S, which partially substitute As, is considered as the cotype (#6260): scanning electron microscopy (SEM), electron microprobe analysis (EMPA) and powder X-ray diffraction (XRD) study were performed for this variety. Additional specimens of the variety chemically close to the holotype but morphologically different were collected from other pockets in the same area in July 2022 (#7529). The temperature measured in the pockets with evseevite during sampling varied from 380 to 450°C.

These chemical varieties of evseevite occur in different mineral assemblages. In holotype and later collected specimen #7529, the new mineral is associated with sanidine, hematite, tenorite, aegirine, cassiterite, sylvite, halite, johillerite, badalovite, calciojohillerite, hatertite, arsmirandite, yurmarinite, axelite, polyarsite, aphthitalite, potassic-magnesio-fluoro-arfvedsonite, litidionite, ferrisanidine, and tridymite. In the cotype specimen, evseevite is associated with hematite, fluorophlogopite, svabite, fluorapatite, tilasite, calciojohillerite, forsterite, cassiterite, belomarinaite, and aphthitalite.

Evseevite occurs in cavities as prismatic, typically long-prismatic to acicular or hair-like crystals up to 0.1 mm, rarely up to 0.7 mm long and up to 0.03 mm thick. They are elongated along [100] and usually combined in parallel, near-parallel, sheaf-, bush- or brush-like aggregates. Clusters (up to 0.5 mm across) of randomly oriented crystals are also common. Acicular crystals form interrupted brushes and hair-like crystals compose pilous crusts (Fig. 1) up to  $2 \times 2$  mm in area. Some crystals are skeletal, case-like.

We suggest that evseevite was deposited directly from the gas phase as a volcanic sublimate or, more probably, formed as a result of the interaction between fumarolic gas and basalt scoria at the temperatures not lower than 450°C. Basalt could be a source of Mg which has low volatility in such fumarolic systems (Symonds and Reed, 1993).

# Physical properties and optical data

Evseevite is transparent, colourless or pale pinkish, with white streak and vitreous lustre. The mineral is brittle, cleavage or parting was not observed. The fracture is uneven. Density calculated using the empirical formula and unit-cell volume found from powder XRD data is  $3.377 \text{ g cm}^{-3}$  for the holotype and  $3.226 \text{ g cm}^{-3}$  for the P- and S-enriched cotype.

The optical data were obtained for the holotype specimen. It is optically biaxial (+),  $\alpha = 1.545(2)$ ,  $\beta = 1.546(2)$ ,  $\gamma = 1.549(2)$ ,  $2V_{\text{meas}} = 40(10)^{\circ}$  and  $2V_{\text{calc}} = 60^{\circ}$  (589 nm). Dispersion of optical axes was not observed. Orientation: X = a. Extinction is straight and elongation is negative. In transmitted plane-polarized light the mineral is colourless and non-pleochroic.

## Raman Spectroscopy

The Raman spectrum of the holotype evseevite (Fig. 2) was obtained on a randomly oriented crystal using an EnSpectr R532 instrument with a green laser (532 nm) at room temperature. The output power of the laser beam was about 10 mW. The spectrum was processed using the EnSpectr expert mode program in the range from 4000 to 100 cm<sup>-1</sup> with the use of a holographic diffraction grating with 1800 lines per cm<sup>-1</sup> and a resolution of 6 cm<sup>-1</sup>. The diameter of the focal spot on the sample was about 10  $\mu$ m. The backscattered Raman signal was collected with a 40x objective; signal acquisition time for a single scan of the spectral range was 1500 ms and the signal was averaged over 20 scans.

The bands in the Raman spectrum of evseevite are assigned according to Nakamoto (1986). Bands in the region between 950 and 770 cm<sup>-1</sup> correspond to As<sup>5+</sup>–O stretching vibrations of AsO<sub>4</sub><sup>3-</sup> anions. The presence of two strong bands in this region (with maxima at 883 and 808 cm<sup>-1</sup>) is caused by significant distortion of AsO<sub>4</sub> tetrahedron (see below). Bands with frequencies lower than 520 cm<sup>-1</sup> correspond to bending vibrations of AsO<sub>4</sub> tetrahedra, Mg–O stretching vibrations and lattice modes. The absence of bands with frequencies higher than 950 cm<sup>-1</sup> indicates the absence of groups with O–H, C–H, C–O, N–H, N–O and B–O bonds in evseevite.

#### **Chemical composition**

The chemical composition of evseevite was studied by EMPA using a Jeol JSM-6480LV scanning electron microscope equipped with an INCA-Wave 500 wavelength-dispersive spectrometer (Laboratory of Analytical Techniques of High Spatial Resolution, Dept. of Petrology, Moscow State University), with an acceleration voltage of 20 kV, a beam current of 20 nA, and a 3  $\mu$ m beam diameter.

The chemical data in wt% are given in Table 1. Contents of other elements with atomic numbers higher than carbon were below detection limits.

The empirical formulae calculated on the basis of O+F = 5 atoms per formula unit are  $(Na_{1.99}Ca_{0.03}K_{0.01})_{\Sigma 2.03}(Mg_{0.98}Fe^{3+}_{0.01}Zn_{0.01}Cu_{0.01})_{\Sigma 1.01}[(As_{0.98}Si_{0.01})_{\Sigma 1.01}O_4](F_{0.97}O_{0.03})$  for holotype and  $Na_{2.02}(Mg_{1.00}Fe^{3+}_{0.03})_{\Sigma 1.03}[(As_{0.69}P_{0.25}S_{0.07})_{\Sigma 1.01}O_4](F_{0.78}O_{0.22})$  for cotype. The idealised formula is  $Na_2Mg(AsO_4)F$  which requires  $Na_2O$  27.16, MgO 17.66,  $As_2O_5$  50.36, F 8.33, -O=F- 3.51, total 100 wt%.

The Gladstone–Dale compatibility index (Mandarino, 1981) for the holotype evseevite,  $1 - (K_p/K_c) = 0.028$ , excellent.

## X-ray crystallography and crystal structure determination details

Single-crystal XRD studies of the holotype sample of evseevite were carried out at room temperature using an Xcalibur S diffractometer equipped with a CCD detector (MoK $\alpha$ -radiation). Powder XRD data for both holotype and cotype samples were collected with a Rigaku R-AXIS Rapid II single-crystal diffractometer equipped with cylindrical image plate detector (radius 127.4 mm) using Debye-Scherrer geometry, CoK $\alpha$  radiation (rotating anode with VariMAX microfocus optics), 40 kV, 15 mA, and exposure 15 min. Angular resolution of the detector is 0.045 2 $\Theta$  (pixel size 0.1 mm). The data were integrated using the software package Osc2Tab (Britvin *et al.*, 2017). Powder XRD data (in Å for CoK $\alpha$ ) for the holotype are given in Table 2. The strongest reflections of powder XRD patterns and unit-cell parameters calculated from powder data for both holotype and cotype are presented in Table 3.

The single-crystal XRD dataset used for the structure model determination was obtained on the same diffractometer Xcalibur S CCD. Data reduction was performed using CrysAlisPro, version 1.171.37.34 (Agilent..., 2014). The data were corrected for Lorentz factor and polarization effects. The crystal structure was solved by direct methods and refined in the space group *Pbcn* using the SHELX software package (Sheldrick, 2015). Low quality of the crystal and consequently of the experimental data precluded excellent agreement between observed and calculated *F* values but resulted in acceptable agreement with  $R_{hkl} = 0.1106$  for 1116 reflections with  $I>2\sigma(I)$ . Reasonable values of interatomic distances and displacement parameters of atoms, as well as good agreement between the measured and calculated powder XRD patterns (Table 2; Fig. 3) showed that the obtained structure model is correct. Further refinement of the structure of holotype evseevite was performed by Rietveld method using this model.

Data treatment and the Rietveld structure analysis were carried out using JANA2006 program package (Petříček *et al.*, 2006). The profiles were modeled using a pseudo-Voigt function. The structure was refined in isotropic approximation of atomic thermal displacements, the values of  $U_{iso}$  for all anions were restricted to be equal. The interatomic distances for Ascentred tetrahedron and Mg-centred octahedron were softly restrained nearby the values obtained for the single-crystal model. Final agreement factors are:  $R_{wp} = 0.0068$ ,  $R_p = 0.0047$ ,  $R_{obs} = 0.0435$ . The observed and calculated powder XRD diagrams demonstrate very good agreement (Fig. 3; Table 2).

Data collection information and structure refinement details for both single-crystal and powder XRD studies are presented in Table 4, coordinates and thermal displacement parameters of atoms are given in Table 5, selected interatomic distances in Table 6, and bond valence calculations in Table 7. The data presented in Tables 5–7 are obtained in the result of the Rietveld refinement.

## Discussion

Evseevite, ideally Na<sub>2</sub>Mg(AsO<sub>4</sub>)F, is the isostructural arsenate analogue of moraskoite Na<sub>2</sub>Mg(PO<sub>4</sub>)F (Karwowski *et al.*, 2015) and synthetic Na<sub>2</sub>Mg(PO<sub>4</sub>)F (Swafford and Holt, 2002). The structure of evseevite (Fig. 4a) is based upon the heteropolyhedral (010) layers formed by [100] chains of Mg-centred octahedra connected *via* AsO<sub>4</sub> tetrahedra (Fig. 4b). A repeat unit of the chain is dimer [Mg<sub>2</sub>O<sub>6</sub>F<sub>2</sub>]<sup>10-</sup> consisting of two MgO<sub>4</sub>F<sub>2</sub> octahedra sharing common face O(2)–F(1)–O(2). Adjacent dimers are linked *via* F(2) vertices. Na cations occupy two crystallographically non-equivalent sites. Na(1) cations centre Na(1)O<sub>4</sub>F<sub>2</sub> octahedra [in moraskoite this site centres seven-fold polyhedron with one elongated Na(1)–O(1) distance of 2.93 Å; in evseevite this distance is more than 3.0 Å and, thus, O(1) is excluded from the coordination sphere of Na(1) cation]. Na(2) cations occupy seven-fold polyhedra Na(2)O<sub>5</sub>F<sub>2</sub>. The Na(1) sites are located between the heteropolyhedral layers whereas the Na(2) sites are situated at both sides of the heteropolyhedral layer.

The crystal structure of evseevite, as well as of moraskoite, can be described in terms of anion-centred polyhedra. Both F(1) and F(2) sites are octahedrally coordinated by two Mg and four Na cations each. F-centred octahedra  $[FNa_4Mg_2]^{7+}$  share faces to form [100] chains  $[FNa_2Mg]^{3+}$ ; AsO<sub>4</sub> tetrahedra are located between the chains (Fig. 5). This approach was used for the description of nacaphite Na<sub>2</sub>Ca(PO<sub>4</sub>)F (Krivovichev *et al.*, 2007), the monoclinic (*P*2<sub>1</sub>/*c*) mineral which is structurally related to moraskoite and evseevite and has the same stoichiometry. The relation of the crystal structures and unit-cell metrics of nacaphite and moraskoite were in detail reported by Karwowski *et al.* (2015).

Evseevite, moraskoite and nacaphite belong to quite few minerals with antiperovskite structure, *i.e.*, have perovskite-type structures but with anions replaced by cations and *vice versa*. Earlier known natural antiperovskites include sulfates, phosphates and silicates (Krivovichev, 2008; Karwowski *et al.*, 2015). Evseevite is the first arsenate mineral with antiperovskite units: its structure is based on F-centred octahedra [FNa<sub>4</sub>Mg<sub>2</sub>]<sup>7+</sup> which form chains similar to those in hexagonal 2*H*-perovskites. We have not found an information on synthetic arsenate antiperovskites in literature and databases, however, the *X*-centred octahedra [*X*Na<sub>6</sub>]<sup>5+</sup> (*X* = F, OH) are present in the structures of synthetic cubic arsenates Na<sub>7</sub>F(AsO<sub>4</sub>)<sub>2</sub>·19H<sub>2</sub>O and Na<sub>7</sub>(OH)(AsO<sub>4</sub>)<sub>2</sub>·19H<sub>2</sub>O. Avdontceva *et al.* (2021) suggested that the presence of the [FNa<sub>6</sub>]<sup>5+</sup> units makes it possible to consider natrophosphate-type compounds as a precursor for the formation of antiperovskite structure motifs based upon anion-centred octahedra.

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Prequipiisned Article

	Holotype (#6328)		Cotype (#6260)				
Constituent	Mean*	Range	Stand. dev.	Mean*	Range	Stand. dev.	Probe standard
Na <sub>2</sub> O	26.67	26.11 - 27.09	0.45	28.83	28.11 - 29.25	0.41	albite
K <sub>2</sub> O	0.23	0.04 - 0.37	0.12	_	_		microcline
CaO	0.66	0.53 - 0.87	0.15	_	_		wollastonite
MgO	17.05	16.34 - 17.83	0.64	18.51	17.72 – 19.31	0.54	chromite
CuO	0.23	0.00 - 0.39	0.17	_	_		Cu
ZnO	0.30	0.21 - 0.38	0.09	_	_		ZnS
Fe <sub>2</sub> O <sub>3</sub> **	0.45	0.34 - 0.57	0.10	1.17	0.37 – 1.53	0.40	magnetite
SiO <sub>2</sub>	0.22	0.16 - 0.30	0.06	_	_	X	wollastonite
P <sub>2</sub> O <sub>5</sub>	_	_		8.17	6.89 - 9.17	0.88	LaPO <sub>4</sub>
As <sub>2</sub> O <sub>5</sub>	48.83	47.97 - 49.39	0.62	36.21	34.12 - 39.51	1.99	InAs
SO <sub>3</sub>	_	_		2.65	0.94 - 4.53	1.55	ZnS
F	8.01	7.78 - 8.21	0.19	6.78	6.31 - 7.55	0.44	fluorophlogopite
–O=F	-3.37			-2.85			
Total	99.28			99.47			

Table 1. Chemical composition (in wt%) of evseevite.

Dash means the content below detection limit; \*averaged for four spot analyses for holotype and for seven analyses for cotype; \*\*trivalent state of admixed iron is assumed because of strongly oxidizing conditions of mineral formation in the Arsenatnaya fumarole: all iron minerals known here contain only Fe<sup>3+</sup> (Pekov *et al.*, 2018a).

$I_{\rm obs},\%$	$d_{\rm obs}$	$I_{\text{calc}}, \%^*$	$d_{ m calc}$	h k l
14	7.05	13	7.063	020
11	6.08	10	6.087	021
2	4.982	4	4.981	110
4	4.599	3	4.600	111
100	4.001	100	4.007	121
5	3.831	6	3.833	112
31	3.527	28,6	3.531, 3.527	040, 130
56	3.479	63	3.482	023
5	3.380	4	3.384	131
5	3.117	5	3.119	113
45	3.041	44	3.044	042
29	2.998	28	3.001	004
2	2.916	1	2.914	123
3	2.851	1	2.858	141
44	2.657	51	2.661	200
68	2.642	10, 64	2.646, 2.642	133, 142
36	2.613	40	2.614	104
6	2.494	8	2.495	150
12	2.437	7.6	2.438.2.433	221, 202
6	2.353	5	2.354	060
4	2.283	3	2.286	134
4	2.274	2.2	2.275. 2.273	231.025
1	2.187	1	2.189	213
3	2.167	2	2.161	232
2	2.101	1	2.125	240
4	2.120	2 1	2.119 2.117	161 153
3	2.099	2, 1	2.101	144
5	2.099	4	2 090	125
4	2.028	4	2.029	063
33	2.002	35.4	2.023 2.001	242 006
16	1 992	15	1 991	204
2	1.971	2	1.972	214
3	1.971	2	1.972	154
4	1.913	4	1.912	251
6	1.915	7	1.912	163
5	1.872	7 Д	1.873	105
3	1.856	3	1.857	116
2	1 844	2	1.843	252
1	1.044	1	1.834	232
1	1 700	1	1.004	172
15	1.75	16	1.000	080
32	1.703	37	1.700	046
5	1.741	57 Л	1./71	321
<u> </u>	1.703	1 2	1.703	262 212
<u>∠</u> 	1.090	1, 2	1.092, 1.089	202, 312
4	1.000	1, 3	1.00/, 1.000	233, 027
4	1.002	3,1	1.000, 1.000	330, 181

Table 2. Powder X-ray diffraction data (d in Å) of evseevite (the holotype).

2	1.644	3	1.645	331
5	1.613	1, 3, 3	1.615, 1.614, 1.611	083, 182, 313
4	1.602	4	1.603	165
3	1.591	1, 2	1.594, 1.590	271, 127
1	1.554	2	1.553	272
12	1.532	4, 13	1.533, 1.533	333, 342
9	1.529	6	1.527	304
5	1.524	1, 3, 1	1.525, 1.522, 1.520	066, 084, 264
3	1.502	4	1.502	350
1	1.493	1	1.491	351
6	1.471	9	1.471	280
9	1.463	12	1.463	184
11	1.457	15	1.457	246
4	1.444	6	1.444	108
1	1.433	1	1.434	217
1	1.429	1	1.429	282
4	1.412	4	1.412	227
3	1.408	1, 3	1.407, 1.407	361, 353
1	1.399	1	1.399	325

\*For the calculated pattern, only reflections with intensities  $\geq 1$  are given. The strongest reflections are marked in boldtype. 

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Mineral	Evseevite	Evseevite	Moraskoite*
	(holotype)	(cotype, P- and S-	
		enriched variety)	
Ideal formula	Na <sub>2</sub> Mg(A	sO <sub>4</sub> )F	Na <sub>2</sub> Mg(PO <sub>4</sub> )F
Empirical	(Na1.99Ca0.03K0.01)52.03	Na <sub>2.02</sub> (Mg <sub>1.00</sub> Fe <sup>3+</sup> 0.03) $\Sigma$ 1.03	Na1.88(Mg1.06Fe <sup>2+</sup> 0.01) $\Sigma$ 1.07
formula	$(Mg_{0.98}Fe^{3+}0.01Zn_{0.01}Cu_{0.01}) \ge 1.01$	$[(As_{0.69}P_{0.25}S_{0.07})\Sigma_{1.01}O_4]$	$[P_{1.00}O_{4.00}]F_{1.00}$
Crystal system	Orthorho	mbic	Orthorhombic
Snace group	Pher	1	Phen
	5 3224(1)	5 307(2)	5 2117(10)
	$14\ 1255(3)$	14.053(3)	13,711(3)
c Å	120047(3)	11.000(3)	11 665(2)
$V Å^3$	902 53(4)	890 5(8)	833 6(3)
Z	8	8	8
Strongest	4.001 - 100	3.984 - 100	3.909 - 75
reflections of	3.527 - 31	3.511 - 26	3.382 - 52
the powder X-	3.479 - 56	3.467 - 48	2.955 - 90
ray diffraction	3.041 - 45	3.027 - 35	2.606 - 100
nattern.	2.998 - 29	2.992 - 27	2.571 - 96
d  Å - I %	2.657 - 44	2.649 - 95	2.547 - 68
<i>w</i> , <i>i i</i> , <i>i</i> , <i>i</i> , <i>i</i> , <i>i</i> , <i>i</i> , <i>i</i>	2.642 - 68	2.631 - 56	1.691 - 67
	2.613 - 36	2.608 - 31	
	2.002 - 33	1.995 - 32	
	1.741 - 32	1.733 - 24	
$D_{\text{calc}}, \text{g cm}^{-3}$	3.337	3.226	2.925
Optical data	Biaxial (+)	none examined	Biaxial
α	1.545(2)		
β	1.546(2)		$n_{\text{mean}} = 1.550(4)^{**}$
γ	1.549(2)		
birefringence	0.004		<i>ca</i> . 0.004
$2V_{\rm meas}$	40(10)°		small
Source	this wo	ork	Karwowski et al.,
			2015

Table 3. Comparative data of evseevite (holotype and cotype) and moraskoite.

\*Powder X-ray diffraction study of moraskoite was not carried out, only the calculated data were reported.

\*\*Probably the mean refractive index for moraskoite was determined wrongly. Our calculation of the Gladstone–Dale compatibility index for moraskoite using  $n_{\text{mean}} = 1.550$  gave  $1 - (K_p/K_c) = -0.079$ , fair. Estimated value of  $n_{\text{mean}}$  for moraskoite calculated using the Gladstone–Dale equation (Mandarino, 1981) is *ca*. 1.51 [1.510 for  $1 - (K_p/K_c) = 0.000$ ].

	Rietveld refinement	Single-crystal data
Ideal formula	Na	<sub>2</sub> Mg(AsO <sub>4</sub> )F
Formula weight		228.2
Temperature, K		293(2)
Radiation and wavelength, Å	CoKα; 1.79021	ΜοΚα; 0.71073
Crystal system, space group, Z	Orthor	hombic, <i>Pbcn</i> , 8
Unit cell dimensions, Å V, Å <sup>3</sup>	a = 5.3224(1) b = 14.1255(3) c = 12.0047(3) 902.53(4)	a = 5.3173(6) b = 14.087(2) c = 12.0110(13) 899.71(19)
Absorption coefficient $\mu$ , mm <sup>-1</sup>	20.042	7.823
F000		864
Diffractometer	Rigaku R-Axis RAPID II	Xcalibur S CCD
Range for data collection, ° Final <i>R</i> indices	$2\theta$ from 9.00 to 140.00 $R_{\rm exp} = 0.0069, R_{\rm wp} = 0.0068,$	$\theta$ from 2.892 to 28.277 -7 $\leq h \leq 7$ , -4 $\leq k \leq 18$ , $0 \leq l \leq 16$ $R_1 = 0.1106$ , w $R_2 = 0.2188 [I > 2\sigma(I)]$ $R_1 = 0.1376$ , w $R_2 = 0.2314$ (all data)
GoF	$R_{\rm p} = 0.0047, R_{\rm obs} = 0.0435$ 0.98	1.30
Largest diff. peak and hole	0.84 and -0.66 e/Å <sup>3</sup>	2.25 and -2.25 e/Å <sup>3</sup>
Other data		Crystal size $0.02 \times 0.05 \times 0.07$ mm
	$\mathbf{Q}^{*}$	Structure solution – direct methods
69		Refinement method – full-matrix least-squares on $F^2$ Absorption correction 'gaussian'

Table 4. Crystal data, data collection information and structure refinement details for evseevite (the holotype).

Site	x	У	Z	$U_{\rm iso}$
As	0.70138(19)	0.88053(11)	0.58705(8)	0.0245(3)
Mg	0.7307(4)	0.51258(18)	0.8186(2)	0.0343(15)
Na(1)	0.7548(7)	0.7389(3)	0.8311(3)	0.0080(13)
Na(2)	0.7413(7)	0.6223(4)	0.5832(2)	0.0308(9)
O(1)	0.8051(9)	0.77571(18)	0.6306(4)	0.0147(6)
O(2)	0.8633(5)	0.9662(2)	0.6536(3)	0.0147(6)
O(3)	0.7442(10)	0.8930(3)	0.45019(13)	0.0147(6)
O(4)	0.3846(3)	0.8969(2)	0.6075(3)	0.0147(6)
F(1)	0.5	0.6201(3)	0.75	0.0147(6)
F(2)	0	0.5923(3)	0.75	0.0147(6)

Table 5. Coordinates and isotropic displacement parameters ( $U_{iso}$ , in Å<sup>2</sup>) of atoms for evseevite (the holotype).

Table 6. Selected interatomic distances (Å) in the structure of evseevite (the holotype).

As - O(1)	1.665(3)	Na(1) - O(1)	2.477(6)
- O(2)	1.687(3)	- O(1)	2.443(6)
- O(3)	1.668(2)	- O(3)	2.349(5)
- O(4)	1.7194(19)	- O(4)	2.465(5)
As – O	< 1.685 >	- F(1)	2.367(5)
		- F(2)	2.634(5)
Mg - O(2)	2.089(4)	Na1 – (O,F)	< 2.456 >
- O(2)	2.145(4)		*
- O(3)	2.071(4)	Na(2) - O(1)	2.266(6)
- O(4)	2.031(4)	- O(2)	2.426(6)
- F(1)	2.120(4)	- O(3)	2.685(7)
- F(2)	2.000(3)	- O(3)	2.715(7)
Mg - (O,F)	< 2.076 >	- O(4)	2.429(5)
,		- F(1)	2.379(3)
		- F(2)	2.467(3)
		Na2 – (O,F)	< 2.894 >

Table 7. Bond valence calculations for evseevite (the holotype).

	As	Mg	Na(1)	Na(2)	Σ
O(1)	1.33		0.16 0.17	0.26	1.92
O(2)	1.25	0.34 0.30		0.18	2.07
O(3)	1.32	0.35	0.21	0.09 0.09	2.06
O(4)	1.14	0.38	0.16	0.17	1.85
F(1)		$0.23^{x2 \rightarrow}$	$0.15^{x2 \rightarrow}$	$0.15^{x2 \rightarrow}$	1.06
F(2)		$0.32^{x2 \rightarrow}$	$0.08^{x2} \rightarrow$	$0.12^{x2 \rightarrow}$	1.04
Σ	5.04	1.92	0.93	1.06	

Bond-valence parameters for As-O, Mg-O and Na-O are taken from Gagné & Hawthorne (2015), and for Mg-F and Na-F from Brese & O'Keeffe (1991).







Fig. 1. Morphology of evseevite: (a-c: holotype, #6328) - aggregates of prismatic to acicular crystals overgrowing badalovite <math>(b - with large hematite crystal); (d-e: cotype, #6260) - long-prismatic to acicular and hair-like crystals on hematite and fluorophlogopite; <math>f - sample #7529, pilous crust completely covering calciojohillerite crystals (with bright white tenorite crystals). SEM images, SE (a, b, d, e) and BSE (c, f) modes.



Fig. 2. The Raman spectrum of evseevite (the holotype).



Fig. 3. Measured and calculated powder X-ray diffraction patterns of evseevite (the holotype). The solid line corresponds to calculated data, the crosses correspond to the measured pattern, vertical bars mark all possible Bragg reflections. The difference between the measured and calculated patterns is shown by curve at the bottom.

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Fig. 5. F-centred octahedra FNa<sub>4</sub>Mg<sub>2</sub> and AsO<sub>4</sub> tetrahedra in the structure of evseevite (two projections). The unit cell is outlined.